

A NEW VERSATILE ROUTE TO 3-HYDROXYPYRROLES¹⁾

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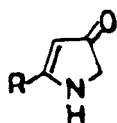
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SUMMARY: N-Acylated 3-hydroxypyrroles 5 have been obtained from a reaction of imides 2 with phosphoranes 3.

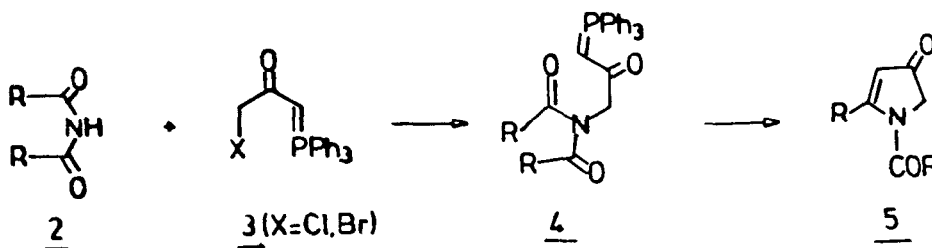
A simple synthesis of N-unsubstituted 1 and N-acylated 3-hydroxypyrroles 5 was imperative for a preparation of 3a-azaazulen-1-ones³⁾, which should be suited also for a preparation of the hitherto unknown parent compound 1a. Derivatives of 1 described so far are usually substituted, at the nitrogen atom and are often difficult to prepare⁴⁾. 5-Methyl-3-hydroxypyrrole 1b, moreover, proved to be extremely labile^{4b)}.

Here we present a simple two-step synthesis of N-acylated 3-hydroxypyrroles 5a-c from symmetric imides 2a-c and the phosphoranes 3⁵⁾ which is depicted the scheme:



1a R=H
1b R=CH₃

	a R=H	b R=CH ₃	c R,R=CH ₂ -CH ₂
<u>4</u>	49%	58%	78%
<u>5</u>	25%	56%	67%



Sodium salts of imides 2 (obtained with sodium hydride at room temperature) were heated with 3 in DMF (4 h, 80°C)⁶⁾. Evaporation and chromatography

gave derivatives 4⁷⁾ which could be cyclized in boiling mesitylene (12 h).

N-Acylated 3-hydroxypyrroles 5⁷⁾ are crystalline and fairly stable compounds, which can be obtained even on a large scale.

Deacylation occurred on treatment of 5b with 5N NaOH in oxygen-free methanol⁸⁾ in a few minutes at room temperature. Isolation and characterisation of 1b followed a procedure described earlier^{4b)}.

ACKNOWLEDGEMENT

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REFERENCES AND NOTES

- 1) We use the name 3-hydroxypyrrole for systematic reasons. Δ^2 -Pyrrolin-4-ones are generally the more stable tautomer, as reflected in the formulae. (Tautomerism of 3-hydroxypyrroles: H. McNab, presented at the 8th Lakeland Heterocyclic Symposium Grasmere 1987).
- 2) 2a) K. Hampel, Dissertation, Univ. Münster 1986.
2b) M. Hohenhorst, Dissertation, Univ. Münster 1986.
- 3) Following communication: W. Flitsch and M. Hohenhorst, *Tetrahedron Lett.*
- 4) a) H. McNab and L.C. Monahan, *J. Chem. Soc. Chem. Commun.*, 1985, 213 and references cited therein.
b) T. Momose, T. Tanaka, T. Yokota, N. Nagamoto and K. Yamada, *Chem. Pharm. Bull.*, 27, 1448 (1979) and references cited therein.
c) For o-substituted 3-hydroxypyrroles, see H.W. Pinnick and K.S. Kochhar *J. Org. Chem.*, 49, 3222 (1984) and references cited therein.
- 5) 3 (X = Cl): R.E. Hudson and P.A. Chopard, *J. Org. Chem.* 28, 2446 (1963).
3 (x = Br): K. Isslab and R. Lindner, *Liebigs Ann. Chem.* 1968, 713.
Yields of the reactions of both phosphoranes do not differ remarkably.
- 6) 2a: 1,2-Dimethoxyethane, 80°C, 8 h.
- 7) The compounds were characterized by elemental analyses as well as by their spectra: selected data: 4a (CDCl₃) δ : 3.3 (br, H, CH=PPh), 4.4 (s, 2H, -CH₂-), 7.5 (m, 15H, phenyl), 9.05 (s, 2H, CHO). 4b (CDCl₃) δ : 2.4 (s, 6H, CH₃), 3.7 (br, 1H, CH = PPh₃), 4.4 (s, 2H, -CH₂-), 7.5 (m, 15H, Aromatic-H). 4c (CDCl₃) δ : 2.6 (s, 4H, CH₂-), 3.7 (br, 1H, CH = PPh₃), 4.25 (s, 2H, -CH₂-), 7.5 (m, 15H, Aromatic-H). 5a (CDCl₃): δ = 4.11, 4.17 (s, 2H, -CH₂-), 5.79, 5.86 (d, J_{21} = 3.97 Hz, 1H, =CH-CO), 8.24, 8.66 (d, J_{12} = 3.97 Hz, N-CH=), 8.50, 8.36 (s, 1H, CHO). 5b (CDCl₃) δ : 2.21 (s, 3H, COCH₃), 2.67 (s, 3H, CH₃), 4.12 (s, 2H, -CH₂-), 5.48 (s, 1H, C=CH-). 5c (CDCl₃) δ : 2.82 (m, 2H, -CH₂-), 3.13 (m, 2H, -CH₂-), 3.97 (s, 2H, -CH₂-), 5.46 (s, 1H, C=CH-).
- 8) Solvents were degassed by ultrasonic treatment and subsequently saturated with argon.

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